

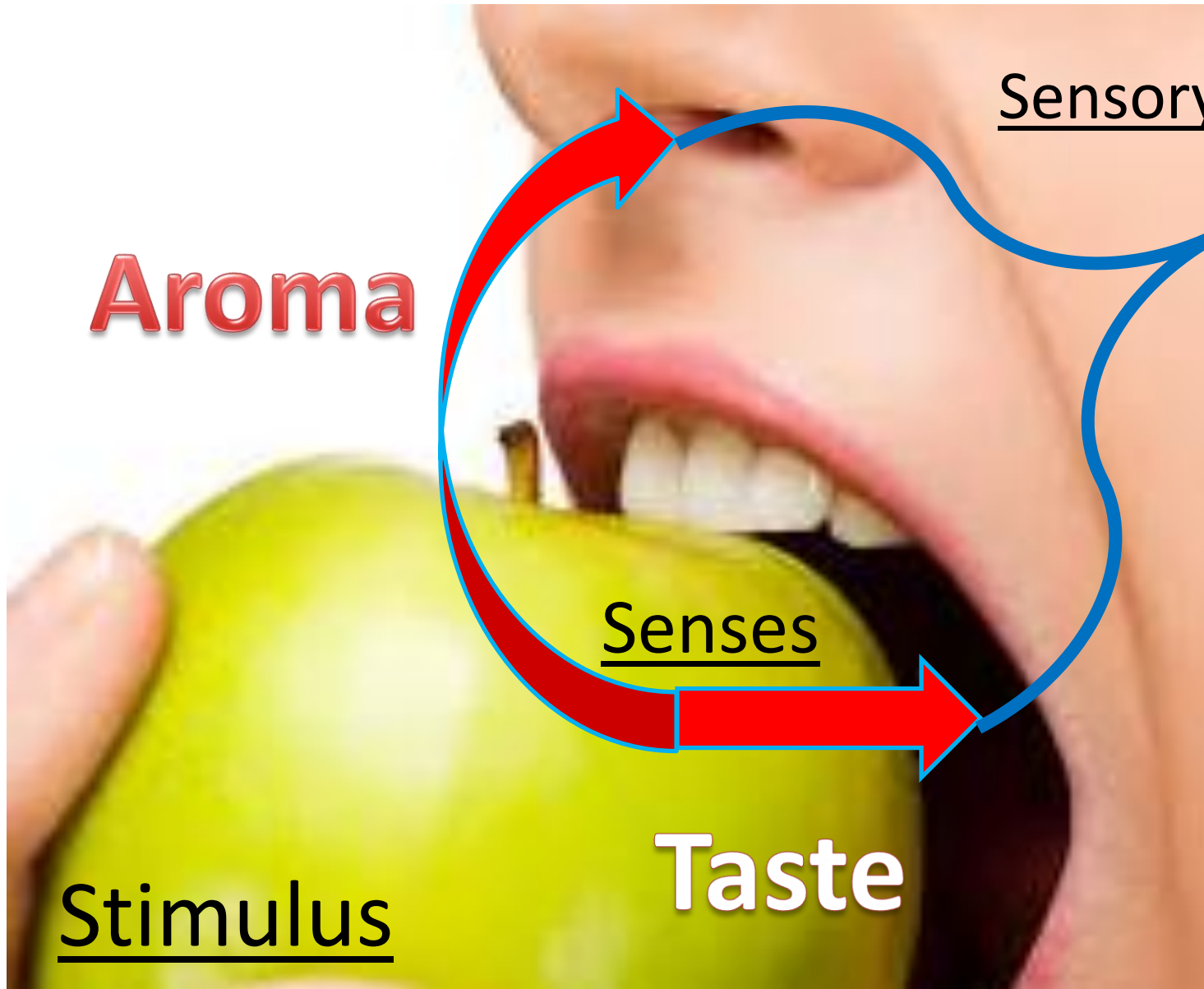
Taste and Aroma in Flavor Sensory



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Indera apa yang digunakan?





Sensory Response

Flavor

Aroma

Senses

Taste

Stimulus



Bahan Pangan



Senyawa



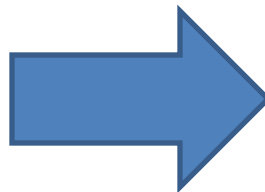
Interaksi dg Reseptor bau&rasa



sinyal



Sistem syaraf pusat



Flavor

mekanisme

Flavor



Senyawa Volatile

Aroma/ Bau

Senyawa Non volatile

Taste/ Rasa

Taste/ Rasa

Manis (sweet)

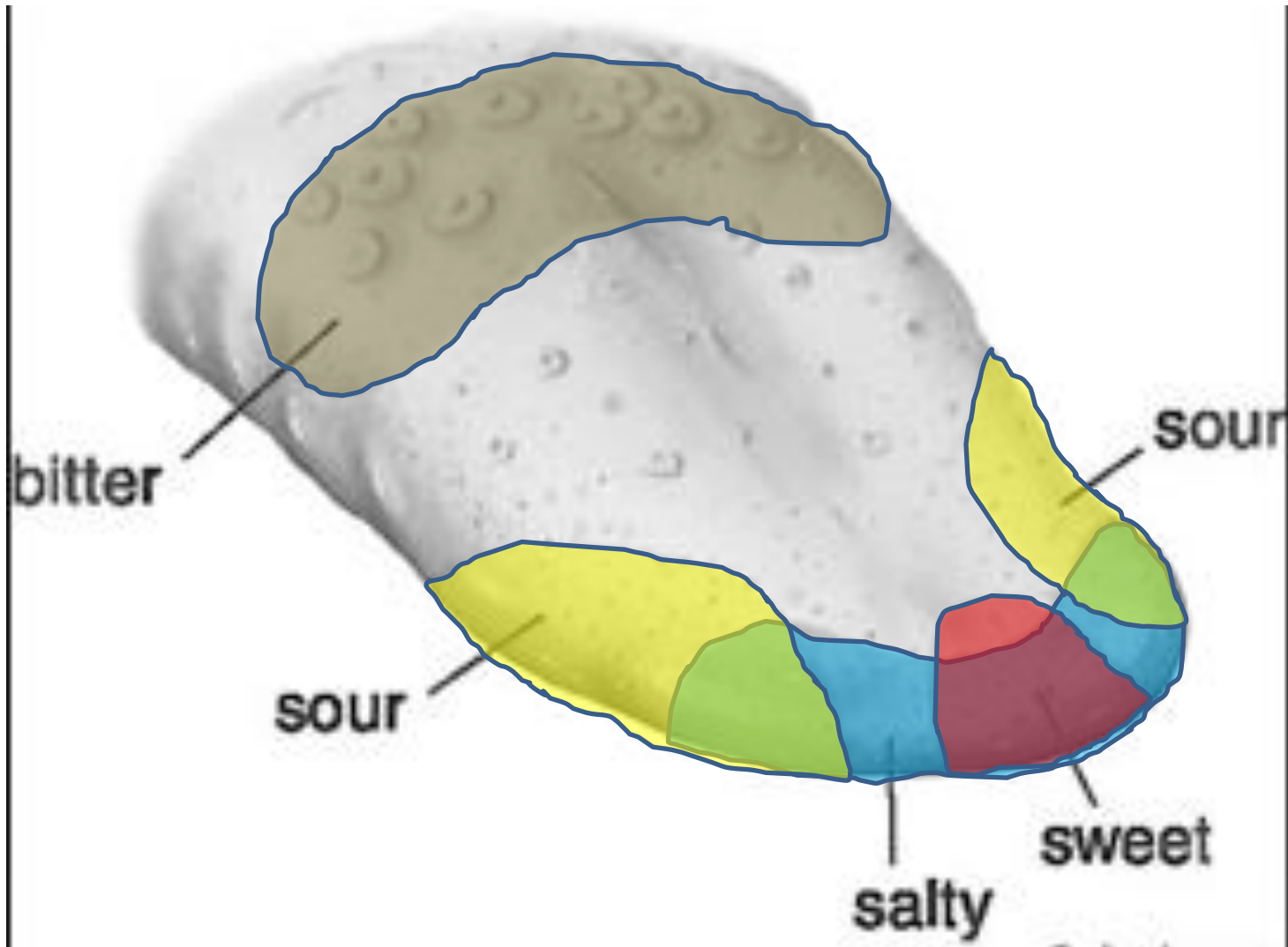
Asam (Acid/ sour)

Pahit (bitter)

Asin (Salty)

Umami (gurih)

Daerah kepekaan rasa di lidah



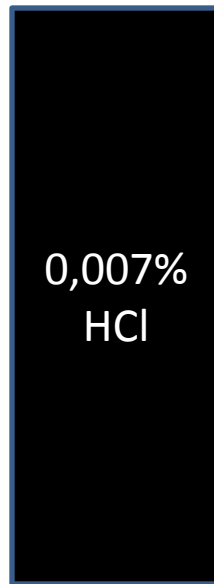
Urutan kepekaan rasa

* Taste threshold

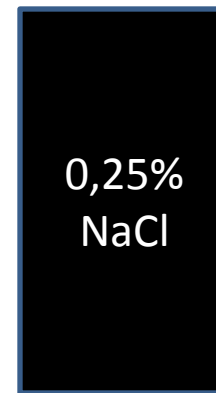
Pahit
(bitter)



Asam
(Acid/ sour)



Asin
(Salty)



Manis
(sweet)



Kualitas rasa yang timbul pada seseorang pada perbedaan konsentrasi NaCl dan KCl

Konsentrasi (mol/l)	NaCl	KCl
0.009	No taste	Sweet
0.01	Faintly sweet	Strongly sweet
0.02	Sweet	Sweet, faintly bitter
0.03	Sweet	Bitter
0.04	Salty, faintly sweet	Bitter
0.05	Salty	Bitter, salty
0.1	Salty	Bitter, salty
0.2	Purely salty	Salty, bitter, sour
1	Purely salty	Salty, bitter, sour

Rasa Manis

Table 4.10. Relative sweetness of sugars and sugar alcohols to sucrose^a

Sugar/ sugar alcohol	Relative sweetness	Sugar/ sugar alcohol	Relative sweetness
Saccharose	100	D-Mannitol	69
Galactitol	41	D-Mannose	59
D-Fructose	114	Raffinose	22
D-Galactose	63	D-Rhamnose	33
D-Glucose	69	D-Sorbitol	51
Invert sugar	95	Xylitol	102
Lactose	39	D-Xylose	67
Maltose	46		
^a 10% aqueous solution.			

Sugar replacer	Function*	Relative sweetness (compared to sugar)^{1,2}	Energy (kcal/g)**
Sugar (Sucrose)	Many	1	4
Acesulfame K	Low-calorie sweeteners	200	0
Aspartame		180-200	4***
Cyclamates		30-50	0
Neotame		7000-13000	0
Saccharin		300-500	0
Glycyrrhizin		30-50	0
Stevia (steviol glycosides)		200-480	0
Sucralose		600	0
Thaumatococin		2000-3000	4***
Inulin	Bulking agents	0.1	2
Polydextrose		0	2
Lactitol	Bulk sweeteners	0.5	2.4
Maltitol		1.0	2.4
Mannitol		0.7	2.4
Sorbitol		0.5-1.0	2.4
Erythritol		0.6-0.8	0
Xylitol		1	2.4
Pectin	Gum/ Thickeners	0	2
Starch		0	4
Guar		0	2

Amino acid

Sweet taste

Glycine

Alanine

Threonine

Proline

Serine

Glutamine

Rasa Masam

Rasa asam dipengaruhi:

Sifat gugus asam

- pH
- Keasaman tertitrasi
- Efek dapar (buffer)
- Adanya senyawa lain (khususnya gula)

Threshold senyawa asam

Salty	equiv/l
<u>HCl</u>	0.0009
Acetic acid	0.0018
Tartaric acid	0.0012
Citric acid	0.0023
<u>Malic acid</u>	0.0005

Sifat beberapa asam, disusun berdasarkan urutan rasa asam yg makin menurun (acuan asam tartrat)

Properties of 0.05N Solutions

<i>Acid</i>	<i>Taste</i>	<i>Total Acid g/L</i>	<i>pH</i>	<i>Ionization Constant</i>	<i>Taste Sensation</i>	<i>Found In</i>
Hydrochloric	+1.43	1.85	1.70	—	—	—
Tartaric	0	3.75	2.45	1.04×10^{-3}	Hard	Grape
Malic	-0.43	3.35	2.65	3.9×10^{-4}	Green	Apple, pear, prune, grape, cherry, apricot
Phosphoric	-1.14	1.65	2.25	7.52×10^{-3}	Intense	Orange, grapefruit
Acetic	-1.14	3.00	2.95	1.75×10^{-5}	Vinegar	—
Lactic	-1.14	4.50	2.60	1.26×10^{-4}	Sour, tart	—
Citric	-1.28	3.50	2.60	8.4×10^{-4}	Fresh	Berries, citrus, pineapple
Propionic	-1.85	3.70	2.90	1.34×10^{-5}	Sour, cheesy	—

Source: From J. Solms, *Nonvolatile Compounds and the Flavor of Foods*, in *Gustation and Olfaction*, G. Ohloff and A.F. Thomas, eds., 1971, Academic Press.

Rasa Asin

Identik dg NaCl

Rasa garam bergantung sifat kation dan anion.

Jika BM kation/ anion / keduanya bertambah besar → berasa Pahit

Table 7–4 Taste Sensations of Salts

<i>Taste</i>	<i>Salts</i>
Salty	LiCl, LiBr, LiI, NaNO ₃ , NaCl, NaBr, NaI, KNO ₃ , KCl
Salty and bitter	KBr, NH ₄ I
Bitter	CsCl, CsBr, KI, MgSO ₄
Sweet	Lead acetate, ¹ beryllium acetate ¹

¹Extremely toxic

Threshold senyawa asin

Salty	Mol/l
NH ₄ Cl	0.004
NaF	0.005
MgCl ₂	0.015
KCl	0.017
NaCl	0.03

Gillette (1985)

NaCl

→ memperkuat rasa mulut (mouthfeel), kemanisan, keseimbangan, & keasinan

→ menutupi/ mengurangi ciri khas menyimpang (off-notes)

Penggantian garam meja yg didasarkan **KCl** =

- tidak memperkuat rasa mulut/ keseimbangan,
- meningkatkan ciri khas pahit/ ciri khas logam (metallic off-notes)

Rasa Pahit

Bisa disebabkan senyawa organik/ anorganik.
Banyak senyawa dari tumbuhan berasa pahit.

Rasa Pahit:

- Beberapa garam anorganik
- Beberapa asam amino
- Senyawa alkaloid
- Senyawa glikosida
- Kuinina
- dll

Threshold senyawa pahit

Bitter	equiv/l
Strychnine hydrochloride	0.000 0016
Quinine	0.000 008
Nicotine	0.000 019
Caffeine	0.000 7
MgSO ₄	0.0046

Table 7-4 Taste Sensations of Salts

<i>Taste</i>	<i>Salts</i>
Salty	LiCl, LiBr, LiI, NaNO ₃ , NaCl, NaBr, NaI, KNO ₃ , KCl
Salty and bitter	KBr, NH ₄ I
Bitter	CsCl, CsBr, KI, MgSO ₄

Garam

Asam amino/ peptida

Bitter taste

Phenylalanine
Tyrosine
Arginine
Leucine
Isoleucine
Valine
Methionine
Histidine

Table 7-5 Taste of Some Selected Peptides

<i>Taste</i>	<i>Composition of Peptides</i>
Flat	L-Lys-L-Glu, L-Phe-L-Phe, Gly-Gly-Gly-Gly
Sour	L-Ala-L-Asp, γ -L-Glu-L-Glu, Gly-L-Asp-L-Ser-Gly
Bitter	L-Leu-L-Leu, L-Arg-L-Pro, L-Val-L-Val-L-Val
Sweet	L-Asp-L-Phe-OMe, L-Asp-L-Met-OMe
Biting	γ -L-Glutamyl-S-(prop-1-enyl)-L-cystein

Source: From J. Solms, *Nonvolatile Compounds and the Flavor of Foods*, in *Gustation and Olfaction*, G. Ohloff and A.F. Thomas, eds., 1971, Academic Press.

Senyawa yg paling dikenal rasa pahit= golongan Alkaloid dan Glikosida

Alkaloid

Senyawa organik yg mengandung Nitrogen & bersifat basa, merupakan turunan dari piridina, pirolidina, kuinolina, isokuinolina atau purina

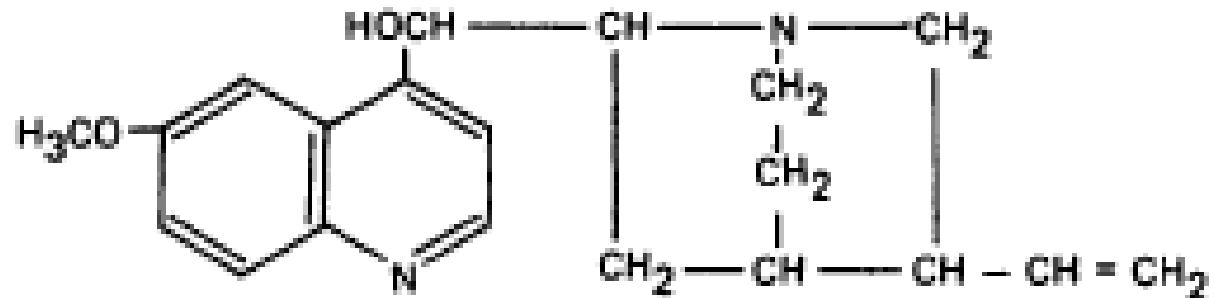
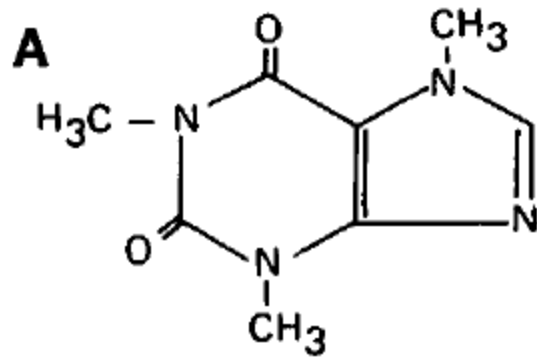


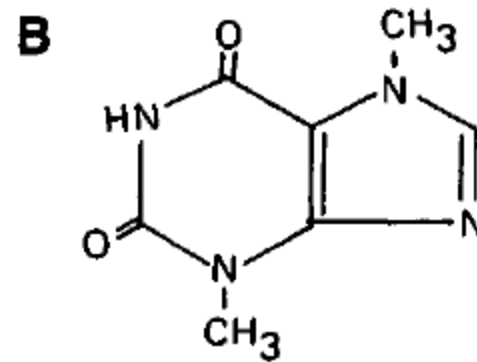
Figure 7-13 Structure of Quinine. This has an intensely bitter taste.

Kuinina → standar pengujian kepahitan.

Kepahitan kuinina hidroklorida dapat dideteksi dalam larutan sangat encer (0,0016% atau 4×10^{-5} mol)



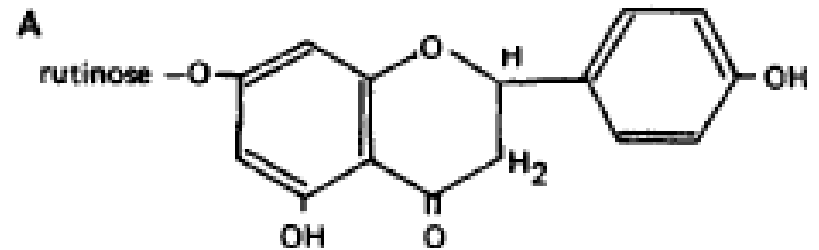
caffeine



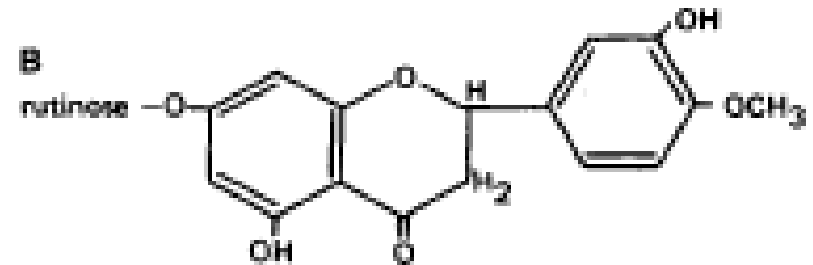
theobromine

Glikosida

Naringin → pada anggur dan jeruk
Lebih pahit daripada quinine
Dapat dideteksi < 0,002%



Hesperidin → citrus, orange, lemon



Umami

(Kawamura and Kare 1987)

Senyawa rasa berbasis asam amino khususnya asam glutamat

Menjadi rasa primer karena:

- Reseptor asam glutamat berbeda dengan rasa manis, asam, asin dan pahit
- Asam glutamat tidak mempengaruhi 4 rasa primer lainnya
- Kualitas rasa asam glutamat berbeda dengan 4 rasa primer lainnya
- Umami tidak dapat diproduksi/ ditiru dari campuran 4 rasa primer lainnya

glutamat

Diproduksi dari :

- Rumput laut
- Gluten gandum
- Limbah gula beet
- Protein kedelai
- Kasein

Table 7–8 Glutamic Acid Content of Some Proteins

<i>Protein Source</i>	<i>Glutamic Acid (%)</i>
Wheat gluten	36.0
Corn gluten	24.5
Zein	36.0
Peanut flour	19.5
Cottonseed flour	17.6
Soybean flour	21.0
Casein	22.0
Rice	24.1
Egg albumin	16.0
Yeast	18.5

Source: From L.A. Hall, Protein Hydrolysates as a Source of Glutamate Flavors, in *Monosodium Glutamate—A Symposium*, 1948, Quartermaster Food and Container Institute for the Armed Forces.

[O-]C(=O)[C@H](N)CC(=O)O.[Na+]

SODIUM SALT OF GLUTAMIC ACID

Abstract



Na Glutamat terdeteksi $<0,03\%$,
pada $0,05\%$ rasanya sangat kuat,
tidak bertambah kuat pada konsentrasi diatas
 $0,05\%$

Rasa glutamat efektif pada pH 6-8.
Menurun pada pH rendah

Aroma/ Odor/ Bau

Mekanisme pembau (olfactory) lebih kompleks dari pada perasa (gustation)

Ada ribuan aroma.

Sensitifitas organ pembau 10.000 kali > dari organ perasa.

Jumlah senyawa volatil dalam makanan:

beef (boiled, cooked)—486

beer—562

butter—257

coffee—790

grape—466

orange—203

tea—541

tomato—387

wine (white)—644

Table 7–14 Primary Odors for Humans and Compounds Eliciting These Odors

<i>Primary Odor</i>	<i>Odor Compounds</i>
Camphoraceous	Borneol, <i>tert</i> -butyl alcohol <i>d</i> -camphor, cineol, pentamethyl ethyl alcohol
Pungent (pedas panas)	Allyl alcohol, cyanogen, formaldehyde, formic acid, methylisothiocyanate
Ethereal	Acetylene, carbon tetrachloride, chloroform, ethylene dichloride, propyl alcohol
Floral	Benzyl acetate, geraniol, α -ionone, phenylethyl alcohol, terpineol
Pepperminty	<i>tert</i> -butylcarbinol, cyclohexanone, menthone, piperitol, 1,1,3-trimethyl-cyclo-5-hexanone
Musky	Androstan-3 α -ol (strong), cyclohexadecanone, ethylene cebacate, 17-methylandrostan-3 α -ol, pentadecanolactone
Putrid (busuk)	Amylmercaptan, cadaverine, hydrogen sulfide, indole (when concentrated, floral when dilute), skatole

Source: From J.E. Amoore et al., The Stereochemical Theory of Odor, *Sci. Am.*, Vol. 210, No. 2, pp. 42–49, 1964.

Table 1.1 Cheddar Cheese Lexicon with Identified References

Term	Definition	Reference
Cooked	Aromatics associated with cooked milk	Nonfat milk heated to 85°C for 30 min
Whey	Aromatics associated with cheddar cheese whey	Fresh cheddar whey
Diacetyl	Aromatics associated with diacetyl	Diacetyl
Milk fat/ lactone	Aromatics associated with milk fat	Fresh coconut meat, heavy cream, δ -dodecalactone
Fruity	Aromatics associated with different fruits	Fresh pineapple, canned pineapple juice
Sulfur	Aromatics associated with sulfurous compounds through water, struck match	Boiled mashed egg, H ₂ S bubbled
Free fatty acid	Aromatics associated with short-chain fatty acids	Butyric acid
Brothy	Aromatics associated with boiled meat or vegetable soup stock	Canned potatoes, commercial low-sodium beef broth cubes, vegetable broth cubes, beef broth cubes
Nutty	Nutlike aromatic associated with different nuts	Lightly toasted unsalted nuts, wheat germ, unsalted wheat crackers
Catty	Aromatics associated with tomcat urine	2 Mercapto-2-methyl-pentan-4-one
Cow/phenolic	Aromatics associated with barns and stock trailers	p-Cresol, bandaids, phenol
Bitter	Fundamental taste sensation elicited by caffeine, quinine	Caffeine (0.08% in water)
Salty	Fundamental taste sensation elicited by salts	Sodium chloride (0.5% in water)
Sweet	Fundamental taste sensation elicited by sugars	Sucrose (5% in water)
Sour	Fundamental taste sensation elicited by acids	Citric acid (0.08% in water)
Umami	Chemical feeling factor elicited by certain peptides and nucleotides	MSG (1% in water)
Prickle	Chemical feeling factor of which the sensation of carbonation on the tongue is typical	Soda water

Source: MaryAnne Drake, Department of Food Science, North Carolina State University, Raleigh, NC. With permission.

Table 2.5 Descriptive Sensory Language for Fluid Milk and Rehydrated SMP

Descriptor	Definition	Reference	Preparation
Cooked/ sulfurous	Aromatic associated with cooked milk	Heated milk	Heat pasteurized skim milk to 85°C for 45 min
Sweet aromatic/ cake mix	Sweet aromatic associated with dairy products	Pillsbury white cake mix Vanillin	Dilute 5 mg of vanillin in skim milk
Cardboard/ wet brown paper	Aromatic associated with cardboard	Wet brown paper or cardboard	Soak 2-cm square of brown paper bag or cardboard in warm water for 30 min
Fatty/fried	Aromatic associated with stale oils or old frier oil	2, 4 decadienal	10 ppb in skim milk
Vitamin	Aromatic associated with vitamin supplements or rubber	Enfamil-liquid Polyvisol vitamins	
Free fatty acid	Aromatic associated with free fatty acids	Feta cheese or butyric acid	Crumbled Feta cheese or 20 ppm butyric acid in skim milk
Metallic/ serum-like	Aromatic associated with rare steak juice	Rare steak juice	
Sweet taste	Basic taste associated with sugars	Sucrose	5% Sucrose solution
Salty	Basic taste associated with salts	NaCl	2% NaCl solution
Sour	Basic taste associated with acids	Citric acid	1% Citric acid solution
Bitter	Basic taste associated with various compounds	Caffeine	0.5% Caffeine solution
Astringent	Drying or puckering of oral tissues	Tea	Soak 6 tea bags in water for 10 min

Source: Adapted from Drake, M.A., Karagul-Yuceer, Y., Cadwallader, K.R., Civille, G.V., and Tong, P.S. 2003. Determination of the sensory attributes of dried milk powders and dairy ingredients. *J. Sensory Stud.* 18: 199–216; Carunchia-Whetstone et al., 2003.

Table 1.2 Descriptors Used for Sensory Characterization of Ultrapasteurized Milk

Aroma	Flavor	Texture	Aftertaste
Cooked	Cooked	Viscosity	Drying
Caramelized	Sweet	Drying	Metallic
Grainy/malty	Caramelized	Chalky	Bitter
Other	Bitter	Lingering	Other
	Metallic		
	Other		

Source: From K.W. Chapman, H.T. Lawless, and K.J. Boor, Quantitative descriptive analysis and principal components analysis for sensory characterization of ultrapasteurized milk, *J. Dairy Sci.* 84: 12, 2001. With permission.

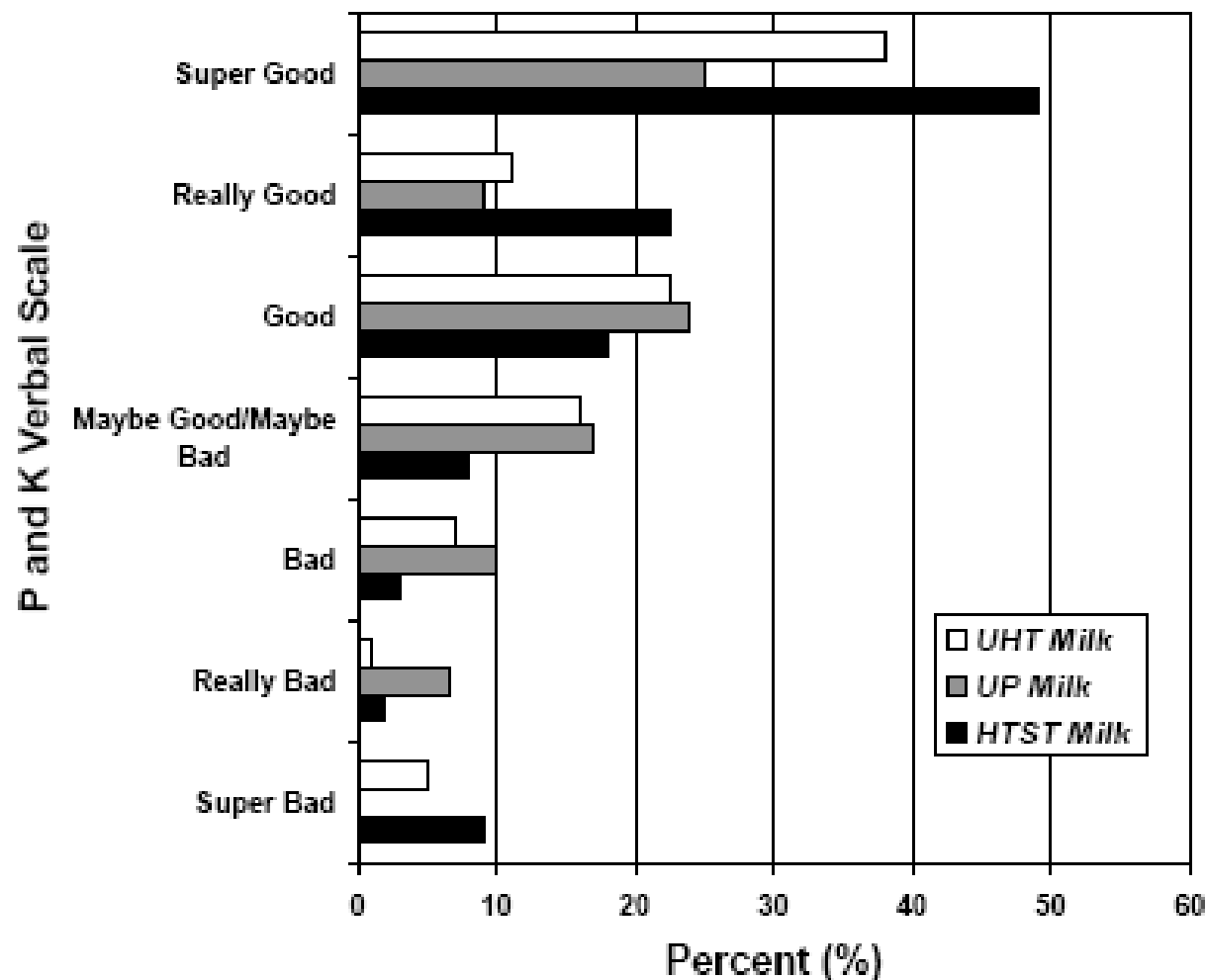


Figure 1.5 The distribution of ratings of milk, using a seven-point hedonic scale. (Black bar represents HTST milk, grey bar represents UP milk, and white bar represents UHT milk.) (From K.W. Chapman and K.J. Boor, Acceptance of 2% ultrapasteurized milk by consumers, 6 to 11 years old, *J. Dairy Sci.* 84: 951, 2001. With permission.)

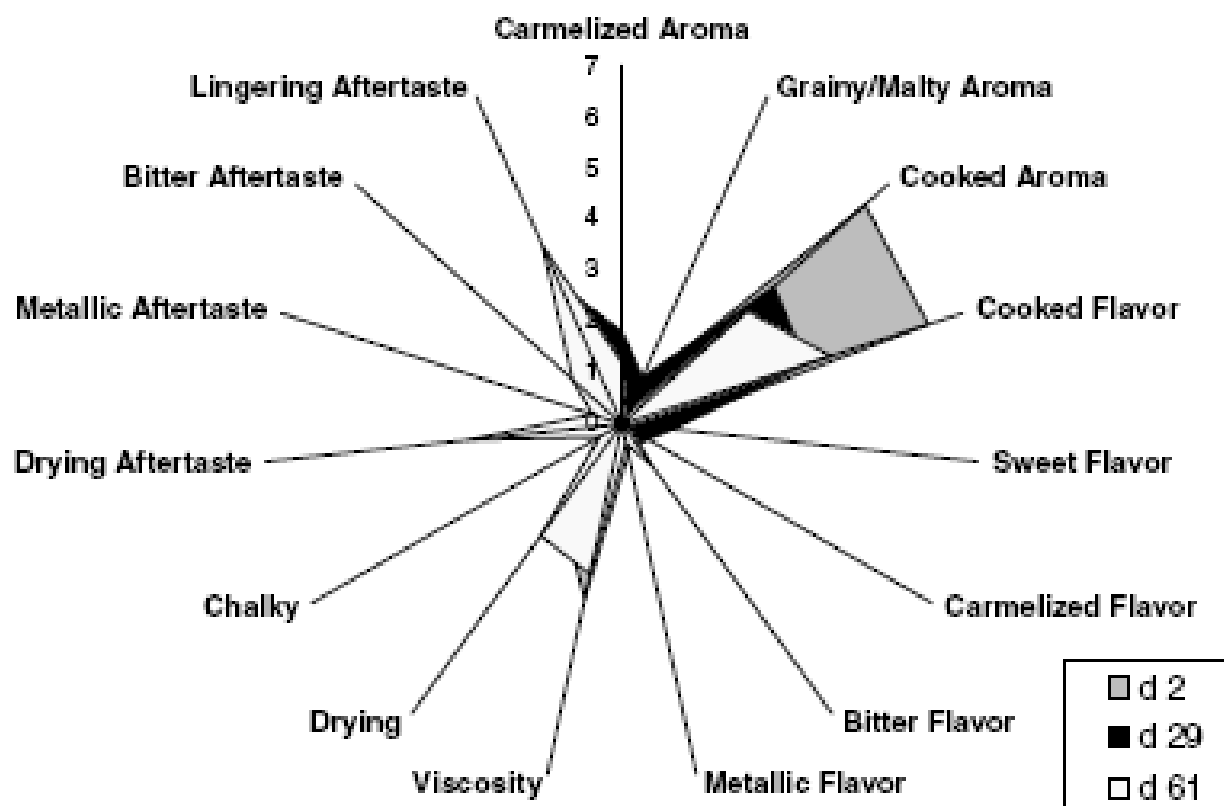


Figure 1.4 Sensory profile for a reduced-fat UP milk sample stored at 6°C for 2 d (dark-grey area), 29 d (black area), and 61 d (light-grey area). (From K.W. Chapman, H.T. Lawless, and K.J. Boor, Quantitative descriptive analysis and principal components analysis for sensory characterization of ultrapasteurized milk, *J. Dairy Sci.* 84: 12, 2001. With permission.)

Tidak semua senyawa volatil berpengaruh terhadap penentuan aroma suatu produk.

Hanya sejumlah sedikit senyawa tertentu dalam produk yang menentukan karakteristik aromanya

TABLE 1. VOLATILE FLAVOURING COMPOUNDS OF DURIAN

Hydrogen sulphide		Methyl acetate	(1)
Methanethiol	*	Ethyl acetate	(3)
Ethanethiol	*	Methyl propionate	(3)
Propanethiol		Ethyl propionate	(1)
Dimethylthioether	*	<i>n</i> -Propyl propionate	(1)
Diethylthioether	*	Ethyl <i>iso</i> -butyrate	(1)
Diethyldisulphide	(1)	Ethyl butyrate	
Methanol	(2)	Methyl α -methylbutyrate	(2)
Ethanol	(5)	Ethyl α -methylbutyrate	(5)
<i>n</i> -Propanol	(4)	<i>n</i> -Propyl α -methylbutyrate	(1)
3-Methylbutan-1-ol		Ethyl <i>iso</i> -valerate	(1)
Acetaldehyde	(1)	Ethyl methacrylate	(1)
Propionaldehyde	(2)	Ethyl benzene	(1)

The relative proportions are based on the height of the GLC peaks as a percentage of recorder full-rate deflection: (1) 10; (2) 10–30; (3) 30–60; (4) 60–100; (5) over 100%. Compounds marked * were identified by TLC in one solvent. All other compounds were identified by their MS obtained from condensed headspace vapours. Identifications were confirmed by comparing MS and chromatograms with those of authentic specimens.



Odor Threshold (OT)



Compounds	Threshold (µg/L)
<i>Alcohols</i>	
1-Pentanol	80,000 ^a
1-Propanol	306,000 ^b
Isobutyl alcohol	40,000 ^c
1-Butanol	150,000 ^d
Isoamyl alcohol	30,000 ^c
4-Methyl-1-pentanol	50,000 ^e
2-Heptanol	70 ^f
1-Hexanol	8000 ^c
(E)-3-hexen-1-ol	400 ^c
(Z)-3-hexen-1-ol	400 ^c
(E)-2-hexen-1-ol	400 ^a
(Z)-2-hexen-1-ol	400 ^a
1-Octan-3-ol	–
1-Heptanol	1000 ^a
2-Ethyl-1-hexanol	8000 ^a
2-Nonanol	58 ^f
Levo-2,3-butanediol	120,000 ^g
Meso-2,3-butanediol	120,000 ^g
1-Octanol	120 ^c
3-(Methylthio)-1-propanol	500 ^c
1-Decanol	400 ^c
Citronellol	100 ^c
Benzyl alcohol	200,000 ^h
2-Phenylethanol	14,000 ^c
1-Dodecanol	1000 ^g
Linalool	25.2 ⁱ
Subtotal	
Proportion (%)	

<i>Esters</i>	
Ethyl acetate	7500 ^c
Isoamyl acetate	30 ^c
Ethyl hexanoate	5 ^c
Ethyl butanoate	–
Hexyl acetate	670 ^b
Ethyl lactate	154,636 ^d
Heptyl acetate	1400 ^a
Methyl octanoate	200 ^j
Ethyl octanoate	2 ^c
Isoamyl hexanoate	–
Ethyl nonanoate	–
Ethyl decanoate	200 ^c
Diethyl succinate	500,000 ^d
Phenethyl acetate	250 ^c
Ethyl dodecanoate	1500 ^a
Subtotal	
Proportion (%)	
<i>Acids</i>	
Acetic acid	200,000 ^c
Propanoic acid	8100 ^h
Isobutyric acid	200,000 ^c
Hexanoic acid	3000 ^c
Heptanoic acid	3000 ^k
Octanoic acid	500 ^c
Decanoic acid	15,000 ^c
Subtotal	
Proportion (%)	
<i>Aldehydes and ketones</i>	
Nonanal	1 ^l
Furfural	14,100 ^c
Decanal	1000 ^e
Benzaldehyde	2000 ^b
Benzylethylaldehyde	–
Geranylacetone	60 ^a

Character impact compound (CIFACs)





Table 1. Integrated and analyzed volatile compounds recovered by solid phase microextraction, gas chromatograph-mass spectrometry in developing and mature cantaloupe, and characteristic impact flavor and aroma compounds [CIFACs (*in italics*)] reported in the *Cucumis melo* literature.²

Compound class		
Compound		
Esters (nonacetate)	Acetates	Aldehydes
methyl 2-methylpropanoate	2-methylpropyl acetate	hexanal
ethyl propanoate	butyl acetate	<i>(E)</i> -2-hexenal
<i>ethyl 2-methylpropanoate</i>	<i>3-methylbutyl acetate</i>	<i>(Z)</i> -3-hexenal
<i>methyl 2-methylbutanoate</i>	<i>2-methylbutyl acetate</i>	benzaldehyde
<i>ethyl butanoate</i>	unknown alkyl acetate	octanal
<i>ethyl 2-methylbutanoate</i>	<i>(Z)</i> -3-hexenyl acetate	benzeneacetaldehyde
ethyl pentanoate	<i>hexyl acetate</i>	<i>(E)</i> -2-octenal
methyl hexanoate	<i>(E)</i> -3-hexenyl acetate	<i>(Z)</i> -6-nonenal
butyl butanoate	<i>(Z)</i> 6-nonenyl acetate	nonanal
<i>ethyl hexanoate</i>	<i>(Z, Z)</i> 3,6-nonadienyl acetate	<i>(E, Z)</i> -2,6-nonadienal
methyl heptanoate	<i>benzyl acetate</i>	<i>(E)</i> -2-nonenal
2-methyl butylbutanoate	phenylethyl acetate	<i>(E, Z)</i> -2,4-nonadienal
ethyl <i>(E)</i> -4-heptenoate	ethyl phenylacetate	<i>(E, E)</i> -2,4-nonadienal
propyl hexanoate		<i>(Z)</i> -citral
ethyl heptanoate		<i>(E)</i> -citral
3-methylbutyl hexanoate		<i>(E)</i> -2-undecanal
pentyl hexanoate		
	Alcohols	Sulfur-compounds and other
	2-methyl 1-butanol	<i>S</i> -methyl thiobutanoate
	<i>eucalyptol</i>	<i>3-(methylthio)propanal</i>
	benzyl alcohol	<i>S</i> -methyl 3-methylbutanethioate
	phenyl ethyl alcohol	<i>ethyl 2-(methylthio)acetate</i>
	<i>(Z)</i> -6-nonenol	<i>methyl 3-(methylthio)propanoate</i>
	<i>(Z, Z)</i> -3,6-nonadienol	<i>ethyl 3-(methylthio)propanoate</i>
	nonanol	<i>3-(methylthio)propyl acetate</i>
	benzenepropanol	<i>(Z)</i> -1,5-octadien-3-one

Odor description

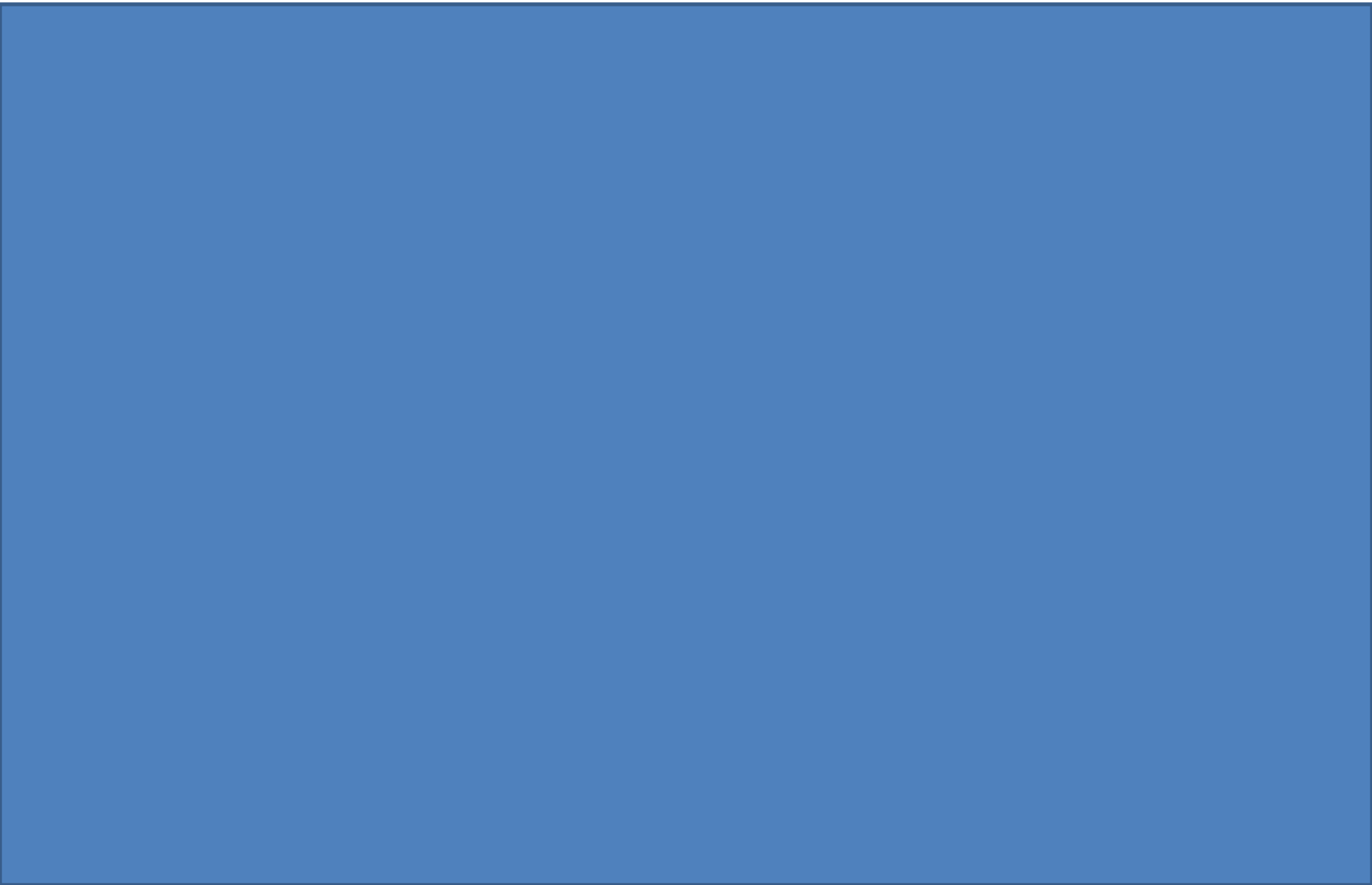


Compound	Odour description ^a
Isoamyl alcohol	Alcoholic
Ethyl hexanoate	Fruity
Unknown	Burnt, plastic
Cis-3-hexen-1-ol	Green plant
Acetic acid	Vinegar
Ethyl octanoate	Fruity
Benzaldehyde	Almond, spicy
Ethyl-3-hydroxy-butanoate	Fruity, grape
Linalool	Floral, rose
Unknown	Chemical, plastic
γ -Butyrolactone	Sweet, caramel
Hotrienol	Floral
Butanoic acid	Chemical, mouldy
Unknown	Floral
α -Terpineol	Floral
Ethyl-4-hydroxy-butanoate	Fruity
Geraniol	Floral
2-Phenylethanol	Floral
Pantolactone	Burnt
4-Vinylguaiacol	Smoky

Odor active value (OAVs)



Aroma reference



Terimakasih

